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## Communications

Band Electronic Structure Study of La<sub>2</sub>CuO<sub>4</sub> and the High-Temperature Superconductor La2-xMxCuO4: Non-Peierls Nature of the Tetragonal to Orthorhombic Distortion of La<sub>2</sub>CuO<sub>4</sub> and Its Implications

Sir:

The report by Bednorz and Müller<sup>1</sup> of possible high-criticaltemperature  $(T_c)$  superconductivity in the Ba-La-Cu-O system touched off intensive studies of the  $La_{2-x}M_xCuO_{4-y}$  phases (M = Ba, Ca, Sr;  $x \le 0.2$ ;  $y \simeq 0$ ),<sup>2-14</sup> eventually leading to the discovery by Wu et al.<sup>15</sup> of the first ambient-pressure superconductor with  $T_c \simeq 94-98$  K, well above liquid-nitrogen temperature (77 K). The structure of the parent compound in this series,  $La_2CuO_4$ , is orthorhombic at room temperature<sup>16</sup> but becomes tetragonal above 533 K.<sup>16b</sup> The orthorhombic structure of La<sub>2</sub>-CuO<sub>4</sub> at room temperature was confirmed in a recent powder neutron diffraction study.<sup>8</sup> Doping La<sub>2</sub>CuO<sub>4</sub> with an alkalineearth metal, M, makes the resulting phase La<sub>2-x</sub>M<sub>x</sub>CuO<sub>4</sub> tetragonal in structure at room temperature, as also shown by the powder neutron diffraction study of La<sub>1.85</sub>Ba<sub>0.15</sub>CuO<sub>4</sub><sup>8</sup> and the single-crystal X-ray diffraction study of  $La_{2-x}Sr_xCuO_4$  (x  $\simeq$ 0.05-0.07).<sup>17</sup> Both the pristine and the doped phases contain

- (1) Bednorz, J. G.; Müller, K. A. Z. Phys. B: Condens. Matter 1986, 64, 189
- Bednorz, J. G.; Takashige, M.; Müller, K. A. Europhys. Lett. 1987, 3, (2) 370
- (3) Takagi, H.; Uchida, S.; Kitazawa, K.; Tanaka, S. Jpn. J. Appl. Phys. Part 2 1987, 26, L123
- Uchida, S.; Takagi, H.; Kitazawa, K.; Tanaka, S. Jpn. J. Appl. Phys., Part 2 1987, 26, L1.
- (5) Cava, R. J.; van Dover, R. B.; Bartlogg, B.; Rietmann, E. A. Phys. Rev. Lett. 1987, 58, 408.
- (6) Chu, C. W.; Hor, P. H.; Meng, R. L.; Gao, L.; Huang, Z. J.; Wang, Y. Q. Phys. Rev. Lett. 1987, 58, 405.
- (7) Chu, C. W.; Hor, P. H.; Meng, R. L.; Gao, L.; Huang, Z. J. Science (Washington, D.C.) 1987, 235, 567.
- Jorgensen, J. D.; Schüttler, H.-B.; Hinks, D. G.; Capone, D. W.; Zhang, (8) K.; Brodsky, M. B.; Scalapino, D. J. Phys. Rev. Lett. 1987, 58, 1024. Tarascon, J. M.; Green, L. H.; McKinnon, W. R.; Hull, G. W.; Gaballe,
- (9) T. H. Science (Washington, D.C.) 1987, 235, 1373
- (10) Bonne, D. A.; Greedan, J. E.; Stager, C. V.; Timusk, T., submitted for publication Solid State Commun.
- (11) Zhao, Z.; Chen, L.; Cui, C.; Huang, Y.; Liu, J.; Chen, G.; Li, S.; Guo, S.; He, Y. Kexue Tongbao (Chin. Ed.), in press.
- (12) Kishio, K.; Kitazawa, K.; Kanbe, S.; Yasuda, I.; Sugii, N.; Takagi, H.; Uchida, S.-I.; Fueki, K.; Tanaka, S. Chem. Lett., in press
- (13) Kanbe, S.; Kishio, K.; Kitazawa, K.; Fueki, K.; Takagi, H.; Tanaka, S., submitted for publication Chem. Lett.
- (14) Kishio, K.; Kitazawa, K.; Sugii, N.; Kanbe, S.; Fueki, K.; Takagi, H.;
- Tanaka, S., submitted for publication Chem. Lett. Wu, M. K.; Ashburn, J. R.; Torng, C. J.; Hor, P. H.; Meng, R. L.; Gao, L.; Huang, Z. J.; Wang, Y. Q.; Chu, C. W. Phys. Rev. Lett. 1987, 58, (15)908.
- (a) Grande, B.; Müller-Buschbaum, Hk.; Schweizer, M. Z. Anorg. Allg. Chem. 1977, 424, 120. (b) Longo, J. M.; Raccah, P. M. J. Solid State (16)Chem. 1973, 6, 526.

layers composed of interlinked CuO<sub>4</sub> moieties, which are constructed from distorted CuO<sub>6</sub> octahedra 1 (with four short and





two long Cu-O distances) upon sharing their "equatorial" oxygen atoms. The CuO<sub>4</sub> layers of  $La_{2-x}M_xCuO_4$  and orthorhombic  $La_2CuO_4$  are flat and bent as depicted in 2 and 3, respectively, and the La<sup>3+</sup> and M<sup>2+</sup> cations are located between the CuO<sub>4</sub> layers. To examine how the dopant M causes high- $T_c$  superconductivity in  $La_{2-x}M_xCuO_4$ , we have carried out tight-binding band calculations<sup>18</sup> on  $La_2CuO_4$  and on a single CuO<sub>4</sub> layer of  $La_{2-x}M_{x}CuO_{4}$ . Our one-electron band electronic structure calculations complement the recent self-consistent-field (SCF) band studies<sup>19,20</sup> on tetragonal La<sub>2</sub>CuO<sub>4</sub>.

As expected from the formal oxidation  $(La^{3+})_2Cu^{2+}(O^{2-})_4$  and the copper coordination of 1, both the SCF band <sup>19,20</sup> and our one-electron band calculations show the  $x^2 - y^2$  band of tetragonal  $La_2CuO_4$  to be half-filled and do not support the suggestion<sup>21</sup> that the  $z^2$  band of La<sub>2</sub>CuO<sub>4</sub> crosses the Fermi level. As far as the d-block bands are concerned, results of three-dimensional (3D) band calculations on La<sub>2</sub>CuO<sub>4</sub> are essentially identical with those

- 6093. (b) Whangbo, M.-H.; Schneemeyer, L. F. *Inorg. Chem.* 1986, 25, 2424. (c) The atomic parameters employed in constructing the extended Hückel<sup>29</sup> Hamiltonian are as follows: The valence shell ionization potential  $H_{ii}$  (eV) and the exponent  $\zeta_i$  of the Slater type atomic orbital xi are respectively -32.3 and 2.275 for O 2s, -14.8 and 2.275 for O 2p, -11.4 and 2.2 for Cu 4s, -6.06 and 2.2 for Cu 4p, -7.67 and 2.14 for La 6s, and -5.01 and 2.08 for La  $6p.^{30}$  The d orbitals of La and Cu are represented by a linear combination of two Slater type orbitals of exponents  $\zeta_i$  and  $\zeta'_i$  with weighting coefficients  $c_i$  and  $c'_i$ , respectively. The  $H_{ii}$  (eV),  $\xi_i$ ,  $c_i$ ,  $\xi_i t'$ , and  $c_i'$  values are respectively -14.0, 5.95, 0.5933, 2.30, and 0.5744 for Cu 3d and -8.21, 3.78, 0.7656, 1.381, and 0.4586 for La 5d.<sup>30</sup> The off-diagonal Hamiltonian matrix elements  $H_{ij}$  were obtained by a modified Wolfsberg-Helmholz formula.3
- Mattheiss, L. F. Phys. Rev. Lett. 1987, 58, 1028. (19)
- Yu, J. J.; Freeman, A. J.; Xu, J.-H. Phys. Rev. Lett. 1987, 58, 1028. Singh, K. K.; Ganguly, P.; Goodenough, J. B. J. Solid State Chem. 1984, 52, 254. (20) (21)

<sup>(17)</sup> Wang, H. H.; Geiser, U.; Thorn, R. J.; Carlson, K. D.; Beno, M. A.; Monaghan, M. R.; Allen, T. J.; Proksch, R. B.; Stupka, D. L.; Kwok, W. K.; Crabtree, G. W.; Williams, J. M. Inorg. Chem. 1987, 26, 1190.
(18) (a) Whangbo, M.-H.; Hoffmann, R. J. Am. Chem. Soc. 1978, 100,



Figure 1. Two-dimensional Fermi surfaces associated with  $x^2 - y^2$  bands of the CuO<sub>4</sub>( $^{6-x}$ )<sup>-</sup> layers present in La<sub>2</sub>CuO<sub>4</sub> and La<sub>2-x</sub>M<sub>2</sub>CuO<sub>4</sub>: (a) the flat CuO<sub>4</sub> $^{6-}$  layer of tetragonal La<sub>2</sub>CuO<sub>4</sub>; (b) the flat CuO<sub>4</sub> $^{5.85-}$  layer of tetragonal La<sub>1.85</sub>M<sub>0.15</sub>CuO<sub>4</sub>; (c) the bent CuO<sub>4</sub><sup>6-</sup> layer of orthorhombic La<sub>2</sub>CuO<sub>4</sub>. In (a) and (b), X and M refer to  $(a^*/2, 0)$  and  $(a^*/2, b^*/2)$ , respectively. Due to the "folded-back" nature of the  $x^2 - y^2$  band in orthorhombic La<sub>2</sub>CuO<sub>4</sub>, which occurs when the unit cell size is doubled, the two Fermi surfaces for the lower and the upper parts of this  $x^2 - y^2$ band are combined into one in the extended Brillouin zone in (c), where **M**, **A**, **B**, and **C** refer to  $(a^*/2, c^*/2)$ ,  $(a^*, 0)$ ,  $(0, c^*)$ , and  $(a^*, c^*)$ , respectively.

of two-dimensional (2D) calculations on a single CuO<sub>4</sub> layer in our one-electron study. Therefore, the d-block bands of  $La_{2-x}M_xCuO_4$  can be approximated by those of its CuO<sub>4</sub> layer, which will be referred to as the  $CuO_4^{(6-x)-}$  layer to indicate the number of electrons in the  $x^2 - y^2$  band (i.e., 1 - x). Shown in 4a-c are the  $x^2 - y^2$  band orbitals of a flat CuO<sub>4</sub><sup>6-</sup> layer 2 for the



wave vectors  $\Gamma = (0, 0)$ ,  $\mathbf{X} = (a^*/2, 0)$ , and  $\mathbf{M} = (a^*/2, b^*/2)$ , respectively. The orbitals of the "axial" oxygen of 1 do not have the correct symmetry to mix into the  $x^2 - y^2$  band, which is therefore dispersionless along the interlayer direction. The p orbitals of the equatorial oxygen atoms make a more effective antibonding interaction with the copper  $x^2 - y^2$  orbital than do the s orbitals of the equatorial oxygen atoms. It is clear from 4 that the  $x^2 - y^2$  band dispersion along  $\Gamma \rightarrow M$  would be about twice as strong as that along  $\Gamma \rightarrow X$ . This dispersion characteristic governs the shape of the Fermi surface (FS)<sup>22</sup> associated with the  $x^2 - y^2$  band.

Shown in Figure 1a is the FS for the half-filled  $x^2 - y^2$  band of a flat  $CuO_4^{6-}$  layer 2. The four pieces of this FS are reasonably well-nested<sup>23</sup> by two wave vectors  $\mathbf{q}_{\pm} = (a^*/2, \pm b^*/2)$ . This nesting is destroyed when electrons are removed from the  $x^2$  –  $y^2$  band as illustrated by the FS of a flat CuO<sub>4</sub><sup>5,85-</sup> layer in Figure 1b. Such observations led to the suggestion<sup>8,19,20</sup> that the tetragonal  $\rightarrow$  orthorhombic distortion of La<sub>2</sub>CuO<sub>4</sub> that occurs at ~533 K<sup>16b</sup> is a Peierls distortion<sup>24</sup> associated with the nesting vector  $\mathbf{q}_{\pm}$ , and an important role of the dopant M is to suppress the Peierls distortion by destroying the FS nesting.

The measured electrical resistivity of orthorhombic La<sub>2</sub>CuO<sub>4</sub> slowly decreases from room temperature to  $\sim 100$  K, then slowly increases from  $\sim 100$  to  $\sim 50$  K, and sharply increases at lower temperatures.<sup>8</sup> This resistivity behavior may be interpreted as characteristic of a doped semiconductor.8 However, the layer bending  $2 \rightarrow 3$ , and the tetragonal  $\rightarrow$  orthorhombic distortion of La<sub>2</sub>CuO<sub>4</sub> as well, cannot open a band gap at the Fermi level due to the presence of twofold screw rotation symmetry along the caxis. As anticipated, our calculations show that orthorhombic  $La_2CuO_4$  is a 2D metal, the FS of which, shown in Figure 1c, is reasonably well nested as in the case of tetragonal  $La_2CuO_4$ . As

suggested earlier,<sup>21,25</sup> therefore, orthorhombic La<sub>2</sub>CuO<sub>4</sub> behaves as a metal at least above  $\sim 100$  K, and thus the tetragonal  $\rightarrow$ orthorhombic distortion is not a Peierls distortion. In fact, our calculations show that the energy of a single  $CuO_4^{(6-x)-}$  layer increases with bending  $(2 \rightarrow 3)$ , although the potential energy curve is very shallow for small bending. Therefore, the driving force for the tetragonal  $\rightarrow$  orthorhombic distortion of La<sub>2</sub>CuO<sub>4</sub>, and that for the layer bending  $2 \rightarrow 3$ , must originate from interactions between  $La^{3+}$  and  $O^{2-}$  ions (from  $CuO_6$  octahedra). Each  $La^{3+}$  (or a combination of  $La^{3+}$  and  $M^{2+}$  in the doped material) is located in a pocket made by four equatorial and four axial oxygen atoms of one  $CuO_4^{6-}$  layer and by one axial oxygen of another  $CuO_4^{6-}$  layer, in which the last axial oxygen is closer to  $La^{3+}$  than to  $Cu^{2+}$ . The interactions between the  $La^{3+}$  and  $O^{2-}$ ions might lead to a soft phonon mode<sup>8,26</sup> responsible for the tetragonal  $\rightarrow$  orthorhombic distortion.

When the tendency for the metal-insulator (MI) transition of a metal is reduced in magnitude, the remnant structural instability may help produce unusually strong electron-phonon coupling<sup>27</sup> and hence high- $T_c$  superconductivity. Consequently, it is crucial to know what causes the "MI" transition of La2CuO4 that occurs below  $\sim 100$  K in understanding how the dopant, M, suppresses it and induces high- $T_c$  superconductivity in  $La_{2-x}M_xCuO_4$ . Provided that the MI transition is not a Mott transition<sup>24c,28</sup> arising from electron-electron repulsion, the structural distortion appropriate for a band gap opening at the Fermi level of orthorhombic La<sub>2</sub>CuO<sub>4</sub> is one that creates two nonequivalent copper atoms (in other words, possible "mixed-valence" formation). One such distortion is the breathing mode displacement of the equatorial oxygen atoms around each copper atom.<sup>19,27c</sup> Further experimental studies on La2CuO4 are necessary to uncover the origin of its MI transition and to help unravel the structural and electronic factors governing the high- $T_c$  superconductivity in  $La_{2-x}M_{x}CuO_{4}$ .

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- (25)
- Ganguly, P.; Rao, C. N. R. Mater. Res. Bull. 1987, 8, 405. Cochran, W. The Dynamics of Atoms in Crystals; Edward Arnold: (26)(a) Matthias, B. T., In Superconductivity; Wallace, P. R., Ed.; Gordon
- (a) Matthias, B. 1., In Superconductivity; Wallace, P. R., Ed.; Gordon and Breach: New York, 1969; Vol. 1 p 227. (b) Hulm, J. K.; Blauger, R. D. In Low Temperature Physics-L713; Timmerhaus, K. D., O'Sullivan, W. J.; Hammel, E. F., Eds.; Plenum: New York, 1974; Vol. 3, p 3. (c) Mattheiss, L. F.; Hamann, D. R. Phys. Rev. B: Condens. Matter 1983, 28, 4227. (d) Whangbo, M.-H.; Williams, J. M.; Schultz, A. J.; Emge, T. J.; Beno, M. A. J. Am. Chem. Soc. 1987, 109, 90. (a) Mott, N. F. Metal-Insulator Transitions; Barnes and Noble: New York, 1977. (b) Brandow, B. H. Adv. Phys. 1977, 26, 651. (c) Hubbard, J. Proc. R. Soc. London. A 1963, 276. 238. (d) Sason S. S.
- (28)bard, J. Proc. R. Soc. London, A 1963, 276, 238. (d) Sason, S. S.; Whangbo, M.-H. Inorg. Chem. 1986, 25, 1201. Hoffmann, R. J. Chem. Phys. 1963, 13, 1397.
- (29)
- (30)
- Alvarez, S., private communication, 1985. Ammeter, J. H.; Bürgi, H. B.; Thibeault, J. C.; Hoffmann, R. J. Am. Chem. Soc. 1978, 100, 3686. (31)

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<sup>(22)</sup> The Fermi surface of a partially filled band is the boundary between the occupied and the unoccupied wave vector regions.

When a piece of a Fermi surface is related to another piece by a (23)translation of wave vector q, the two pieces are said to be nested by q.

<sup>(</sup>a) Peierls, R. E. Quantum Theory of Solids; Oxford University Press: (24)London, 1955; p 108. (b) Berlinsky, A. J. Contemp. Phys. 1976, 17, 331. (c) Whangbo, M.-H. Acc. Chem. Res. 1983, 16, 95. (d) Moret, R.; Pouget, J. P. In Crystal Chemistry and Properties of Materials with Quasi-One-Dimensional Structures; Rouxel, J., Ed.; Reidel: Dordrecht, The Netherlands, 1986; p 87.